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- (A) Process for inhibiting corrosion in oil production fluids.
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#### Description

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Corrosion inhibitors used in oil production offshore are highly cationic but the use of such cationic based corrosion inhibitors for offshore oil platforms are becoming less acceptable for environmental reasons. By being cationic they are attracted to metal surfaces, controlling acid type corrosion. When these cationic corrosion inhibitors find their way into the seawater, they are attracted to a particular type of algae, diatomes. These algae are a part of a food-chain for mussels.

It has been reported that a corrosion inhibitor has not been found, that does not inhibit growth of these algae, at a concentration greater than 1 ppm. The diatomes have a skeleton structure of about 80 % silicon dioxide or quartz. Cationic materials, which make up most corrosion inhibitors are attracted to quartz as readily as they are to metals. Cationics used for producing corrosion inhibitors, can be quaternarry amines, amine salts, ethoxilated amines, ether amins, polyamines, amido amines, essentially all nitrogen based molecules are candidates for producing corrosion inhibitors.

US-A-3,054 750 discloses the use of compounds according to formula CH<sub>19</sub> H<sub>29</sub> CH<sub>2</sub> NHCH<sub>2</sub> CH<sub>2</sub> COOH as corrosion inhibitors in oil well fluids. Amounts of inhibitor in the oil well fluid are within the range of 50 to 400 ppm.

US-A-4,201,678 discloses the use as foaming agent of amphoteric betaine of the formula

$$R^{2}$$
|
 $R^{1}$ -CONH-(CH<sub>2</sub>)<sub>3</sub>- $\Theta$ N-CH<sub>2</sub>-COO<sup>6</sup> (1)
|
 $R^{3}$ 

in combination with a sodium olefin sulfonate in an oil production fluid (drilling fluid). The amount of such additive in the drilling fluid is ranging from 0.08 to 1.0 wt% (800 to 10.000 ppm).

Due to the decreased toxic properties and the ampholytic characteristics of the ampholytes and betaines, the biodegradability is greatly increased. Thereby reducing the possibility of these products ever reaching toxic levels.

It has now been found that this problem can be overcome by using certain betaines or ampholytes as corrosion inhibitors.

Accordingly the subject matter of the instant invention is a process for inhibiting corrosion in oil production fluids which process consists in adding to the oil production fluids under acidic conditions an effective amount of a betaine or ampholyte of the formula 1

$$R^{1}$$
-CONH-(CH<sub>2</sub>)<sub>3</sub>- $\theta$ N-CH<sub>2</sub>-COO $\theta$  (1)

wherein  $R^1$  is  $C_{10}$ - $C_{20}$ -alkyl or  $C_{10}$ - $C_{20}$ -alkenyl, preferably  $C_{14}$ - $C_{18}$ -alkyl or  $C_{14}$ - $C_{18}$ -alkenyl and  $R^2$  and  $R^3$  are  $C_1$ - $C_4$ -alkyl, preferably methyl.

These betaines and ampholytes as described before can be used as such or they can be used upon being neutralized with acids such as, but not limited to acetic acid, adipic acid, sebacic acid, naphthenic acids, paraffinic acids, tall oil acids or free SO<sub>2</sub>. They function as corrosion inhibitors in oil production fluids containing acid such as carbon dioxide as corrodent. Carbon dioxide is the most common acid in oil production fluids.

In addition these betaines and ampholytes can also be used together with other corrosion inhibitors to reduce the toxicity, preferably oxalkylated fatty amines of the formulas 3 or 4

$$R^1 - NH - (C_2 H_4 O)_x H$$
 (3)

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# $R^1NHCH_2CH_2CH_2NH-(C_2H_4O)_xH$ (4)

wherein R1 is as defined above and X is a number from 5 to 15 or a compound of the formula 5

HO  $\stackrel{O}{\longrightarrow} \stackrel{N}{\stackrel{||}{\cup}} = C1$  (5)

The betaine or ampholyte compound of formula 1 or the mixtures of these compounds with the compounds of formulas 3, 4 and/or 5 are added to the oil production fluids at a rate that is effective to prevent corrosion. Under usual conditions an amount of 5 to 300, preferably 2 to 20 ppm will be sufficient.

When the betaines or ampholytes of formula 1 are dumped to neutral or alkaline pH-water, such as seawater, they loose their cationic characteristics, and take on the characteristics of nonionic or anionic molecules. Under these conditions, they no longer function as corrosion inhibitors and they also will not inhibit the growth of the diatomes, as would be the case if they maintained their cationic characteristics.

Also the oxalkylated amines of formulas 3 and 4 do not function as cationics and therefore they loose their ability to be readily absorbed by the silicate in the diatomes. The same is with the p-hydroxy-benzoic acid derivative of formula 5 which hydrolyses at a pH greater than 7.5 forming non-toxic benzoates.

Formulations for use on the basis of the above described betaines and their mixtures with compounds of formulas 3, 4 and/or 5 can be made by dissolving these compounds in a mixture of water and lower alcohols.

### 25 Example 1

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12 % tallow fatty acid-amidopropyl-N,N-dimethyl-N-carboximethyl-betaine 5 % adipic acid 10 % isobutanol 5 % methanol water ad 100 %

# Example 2

12 % tallow fatty acid-amidopropyl-N,N-dimethyl-N-carboximethyl-betaine
 3 % SO<sub>2</sub>
 10 % isobutanol
 5 % methanol
 water ad 100 %

#### Example 3

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8 % tallow fatty acid-amidopropyl-N,N-dimethyl-N-carboximethyl-betaine 10 % isobutanol 5 % methanol

15 % mixture of alkyl pyridines 5 %  $SO_2$  water ad 100 %

## 50 Example 4

10 % tallow fatty acid-amidopropyl-N,N-dimethyl-N-carboximethyl-betaine 5 % cocosdiamine + 15 moles ethylene oxide 10 % isobutanol 5 % methanol water ad 100 %

#### Example 5

10 % tallow fatty acid-amidopropyl-N,N-dimethyl-N-carboximethyl-betaine 5 % compound of formula 5

85 % mixture of propylene glycol and isobutanol

In the examples described before all percentages are by weight.

These compositions have been checked in a standard corrosion inhibitor test, referred to as a bubble test. In the presence of 80 % brine and 20 % crude oil, saturated with carbon dioxide, in a 24 hour test gave greater than 90 % protection, when measured by Corrator, at a treatment rate of 20 ppm. The same products in a dynamic autoclave test with 3 % sodium chloride, 10 bars of carbon dioxide pressure, 12 °C, containing 8 steel coupons, rotating at about 3 meter per second gave 60 % protection. There are a number of commercial inhibitors on the market, that give 40 % or less protection, others give greater than 90 % protection. Therefore, the compositions as described above are within the range of commercial corrosion inhibitors. These compositions have also been tested for growth inhibition of Skeletonrema Costatum, a standard toxicity test for marine diatomes, at a treating rate of 4 ppm and there was no appearent retardation of growth of the Skeletonrema Costatum up to 4 ppm and some at 8 ppm.

#### Claims

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A process for inhibiting corrosion in oil production fluids which process consists in adding to the oil
production fluids under acidic conditions an effective amount of a betaine or ampholyte of the formula
(1)

$$R^{2}$$
|
 $R^{1}$ -CONH-(CH<sub>2</sub>)<sub>3</sub>- $\Theta$ N-CH<sub>2</sub>-COO<sup>O</sup> (1)
|
 $R^{3}$ 

wherein R1 is C10-C20-alkyl or C10-C20-alkenyl and R2 and R3 are C1-C4-alkyl.

- A process as claimed in claim 1, wherein R¹ is C<sub>14</sub>-C<sub>18</sub>-alkyl or C<sub>14</sub>-C<sub>18</sub>-alkenyl and R² og R³ are methyl.
- 3. A process as claimed in claim 1 or 2 which consists in adding 5 to 300, preferably 2 to 20 ppm of the betaine or ampholyte of formula (1) to the oil production fluids.
  - 4. A process as claimed in any of claims 1-3 which consists in using the betaine or ampholyte of formula (1) together with oxalkylated fatty amines of formulas (3) or (4)

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$$R^1 - NH - (C_2H_4O)_xH$$
 (3)

 $R^1$ -NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH(C<sub>2</sub>H<sub>4</sub>O)<sub>x</sub>H (4)

wherein R<sup>1</sup> is as defined for formula 1 and x is a number from 5 to 15, and/or a compound for the formula (5)

HO 
$$\stackrel{O}{=}$$
  $\stackrel{N}{=}$   $\stackrel{OH}{=}$   $\stackrel{C}{=}$   $\stackrel{C}{=}$   $\stackrel{C}{=}$   $\stackrel{C1}{=}$  (5)

## Patentansprüche

 Verfahren zur Korrosionshemmung in Erdölförderflüssigkeiten, welches darin besteht, daß den Erdölförderflüssigkeiten unter sauren Bedingungen eine wirksame Menge eines Betains oder eines Ampholyten der allgemeinen Formel (1) zugesetzt wird:

$$\begin{array}{c|c}
R^{2} \\
 & \\
R^{1}-CONH-(CH_{2})_{3}-{}^{+}N - CH_{2}-COO^{-} \\
 & \\
R^{3}
\end{array} (1)$$

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wobei R¹ ein Alkyl oder Alkenyl mit 10 bis 20 Kohlenstoffatomen und R² und R³ jeweils ein Alkyl mit 1 bis vier Kohlenstoffatomen sind.

- Verfahren nach Anspruch 1, bei welchem R¹ ein Alkyl oder ein Alkenyl mit 14 bis 18 Kohlenstoffatomen und R² und R³ Methyl sind.
  - 3. Verfahren nach Anspruch 1 oder 2, welches darin besteht, den Erdölförderflüssigkeiten 5 bis 300, vorzugsweise 2 bis 20 ppm Betein oder den Ampholyten der allgemeinen Formel (1) zuzusetzen.

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- Verfahren nach einem der Ansprüche 1 bis 3, welches darin besteht, das Betain bzw. den Ampholyten der allgemeinen Formel (1) zusammen mit oxalkylierten Fettaminen der allgemeinen Formel (3) bzw.
   (4)
- 30 R1 NH (CH2H4O)xH (3

 $R^1$ -NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH(C<sub>2</sub>H<sub>4</sub>O)<sub>x</sub>H (4)

wobei R¹ die für die allgemeine Formel (1) angegebene Bedeutung hat und x eine Zahl zwischen 5 und 15 ist, und/oder eine Verbindung der allgemeinen Formel (5):

einzusetzen.

## 45 Revendications

 Procédé pour inhiber la corrosion dans les fluides de production de pétrole, qui consiste à ajouter aux fluides de production de pétrole, en milieu acide, une quantité efficace d'une bétaïne ou un ampholyte répondant à la formule (1):

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$$R^{2}$$

|
 $R^{1}$ -CONH-(CH<sub>2</sub>)<sub>3</sub>- $\Theta$ N-CH<sub>2</sub>-COO<sup>6</sup> (1)
|
 $R^{3}$ 

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dans laquelle  $R^1$  est un groupe alkyle en  $C_{10}$  à  $C_{20}$  ou alcényle en  $C_{10}$  à  $C_{20}$ , et  $R^2$  et  $R^3$  sont des groupes alkyle en  $C_1$  à  $C_4$ .

- 2. Procédé selon la revendication 1, dans lequel R¹ est un groupe alkyle en C₁₄ à C₁ଃ ou alcényle en C₁₄ à C₁ଃ, et R² et R³ sont des groupes méthyle.
  - 3. Procédé selon les revendications 1 ou 2, qui consiste à ajouter 5 à 300, de préférence 2 à 20 ppm de la bétaine ou de l'ampholyte répondant à la formule 1 aux fluides de production de pétrole.

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- 4. Procédé selon l'une quelconque des revendications 1 à 3, qui consiste à utiliser la bétaïne ou l'ampholyte répondant à la formule (1) en même temps que des amines grasses oxalkylées répondant aux formules (3) ou (4)
- 25 R1 NH (C2H4O)xH (3)

 $R^1$ -NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH(C<sub>2</sub>H<sub>4</sub>O)<sub>x</sub>H (4)

dans lesquelles R¹ est tel que défini pour la formule 1 et x est un nombre de 5 à 15, et/ou un composé répondant à la formule (5)

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